Cerium-Catalyzed Reaction of β-Dicarbonyl Compounds with Styrene and Atmospheric Oxygen

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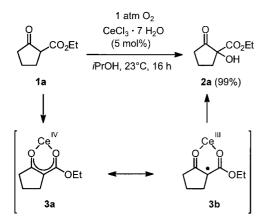
The cerium-catalyzed C–C coupling reaction of carbo- and heterocyclic β -dicarbonyl compounds 1 with styrenes 4, using oxygen (air) as the oxidant, at ambient temperature, is reported. The reaction afforded a mixture of diastereoisomeric hydroperoxides 5 which could be transformed into the

unique dioxo compounds **6** by disproportionation of the peroxo function under treatment with acetyl chloride/pyridine.

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Introduction

Atmospheric oxygen (air) represents the oxidant of choice from an economical and ecological point-of-view, and this makes procedures using oxygen highly attractive. For example, we recently published a novel cerium-catalyzed α -hydroxylation of cyclic β -dicarbonyl compounds 1 with O_2 (Scheme 1). For this reaction we assume a Ce^{IV} diketonate complex 3a to be an intermediate, which may, in the sense of a ligand-to-metal charge transfer, also be regarded as a Ce^{III} complex 3b with a radical β -diketon- α -yl ligand. For the system shown in Scheme 1 with catalytic amounts of the cerium salt, a re-oxidation of Ce^{III} to Ce^{IV} by O_2 seems to be reasonable. $^{[2]}$



Scheme 1. Cerium-catalyzed α -hydroxylation of β -dicarbonyl compound 1a proceeding via the assumed intermediate 3a/3b

Of course, it has been known for a long time that β -dicarbonyl compounds can be oxidized to their α -radicals with stoichiometric amounts of Ce^{IV} or Mn^{III} reagents such as CAN or Mn(OAc)3, and these radicals have been investigated intensively in C-C bond forming reactions. [3] For example, CAN has been utilized for oxidative dimerizations of malonates, [4] as well as for inter-[5] and intramolecular [6] conversions of β-dicarbonyl compounds with olefins. Analogously, α-radicals generated from β-dicarbonyls and Mn(OAc)₃ react with olefins to give lactones and lactams.^[7] Intramolecular insertion of these radicals has been applied to the preparation of various β-lactams, [8] higher lactams, and spirolactams.^[9] The Mn(OAc)₃-mediated reaction of βoxo esters with enol ethers is of particular interest, since the products bear a 1,4-dicarbonyl motif.^[10] This structural motif is commonly accessible only by the application of an umpolung strategy or the transformation of furan derivatives, and is useful in the preparation of annulated cyclopentenone derivatives.[11] All the literature referenced so far has used stoichiometric amounts of Ce^{IV} or Mn^{III} reagents. The re-oxidation of a Mn- or Ce-catalyst with molecular oxygen under the reaction conditions seems to be a highly attractive goal in terms of atom economy. And indeed, the catalytic reaction of \beta-dicarbonyl compounds with olefins and O2 in the presence of sub-stoichiometric amounts of Mn(OAc)₃ has been reported. Endoperoxidic products (1,2dioxane derivatives) are obtained in this reaction as the result of a formal [2+2+2]-cycloaddition process.^[12] However, the yields of these products only reach an acceptable level in the cases of disubstituted olefins. Optimal results are achieved using 1,1-diaryl-substituted alkenes, whereas simple olefins like styrene give low yields.^[13] Applying a Mn^{II}-Co^{II} catalyst system, though, gives satisfying results in the reaction of β-dicarbonyl compounds with simple alkenes and oxygen.^[14] In this case, almost no by-products with oxygen incorporation (for example, as hydroxy or peroxo functionalities) are formed. The use of catalytic

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amounts of cerium compounds and O_2 as the re-oxidizing agent has been reported so far in only one case, resulting, in the cases of monosubstituted olefins such as styrene, once again, in low yields of dihydrofuran products. Herein, we wish to report our results on C-C coupling reactions of styrenes 4 and cyclic and heterocyclic β -dicarbonyl compounds 1 with air as the oxidant, using only a catalytic amount of the cerium salt $CeCl_3$ - $7H_2O$. This reaction gives endoperoxidic 1,2-dioxane derivatives as products, and is analogous to the process using stoichiometric amounts of Mn^{III} and Ce^{IV} .

Results and Discussion

Cyclic β -oxo esters $1\mathbf{a} - \mathbf{c}$ reacted with styrene (4a) and α -methylstyrene (4b) in the presence of a cerium-catalyst and atmospheric oxygen, to give the C-C coupling products $5\mathbf{a} - \mathbf{f}$ and $6\mathbf{a}$, \mathbf{c} (Scheme 2, Table 1).

Scheme 2. Cerium-catalyzed C–C coupling reaction of β -dicarbonyl compounds 1a-1e and styrenes 4a,b with oxygen as oxidant; reagents and conditions: (a) 1 (1 equiv.), 4 (2 equiv.), $CeCl_3 \cdot 7H_2O$ (0.05 equiv.), iPrOH, 23 °C, 16 h, air; for details see Table 1

The reaction conditions were optimized in the reaction of 1a with both 4a and 4b: to suppress the formation of the α -hydroxylated product 2 in favour of the C-C coupling reaction, the conversion was performed under atmospheric oxygen instead of 1 atm O_2 and a twofold excess of the olefin, while again, 2-propanol was found to be the optimal solvent

In the same manner, the five-membered heterocyclic β -dicarbonyl compounds 1d,e reacted with styrenes 4a,b to give the addition products 5g-5j and 6g,i (Scheme 2). The results are listed in Table 1.

In nearly all cases, the donors 1 were completely consumed in the reaction. The reaction of 1 and styrene (4a) afforded the α -hydroxylation products 2 and the dioxo derivatives 6 as by-products. The reaction of the precursor lactam 1e, however, gave the dioxo lactam 6i as the major product in 45 % yield, as well as the C-C coupling product 5i (5 %) (Table 1).

As depicted in Scheme 2, the hydroperoxides 5 exist in their endoperoxidic hemiketal forms; monocyclic hydroperoxides could not be detected. Four diastereoisomers may be expected for three stereocenters, and indeed, in case of the seven-membered ring product 5f, four diastereomers were detected. However, with the exceptions of 5e, 5h (three diastereoisomers) and 5d, 5i (one diastereomer), in general, two diastereoisomers were observed. In case of the lactonederived hydroperoxides 5g and 5h, the two diastereoisomers could be separated by column chromatography on silica gel. While the diastereomers of 5g were obtained in a 1:1 ratio, two diastereomers of 5h were isolated in a ratio of 7:1 with a trans-configured isomer as the major product, as proved by the X-ray crystal structure. Crystallization of both diastereoisomers of 5g and 5h from dichloromethane gave single crystals which were suitable for X-ray crystallographic analysis (Figure 1 and Figure 2).[16]

The relative configuration of both diastereomers of **5g** demonstrates a C-C bond-formation without any stereoselectivity: the lactone and the phenyl group are arranged either in the relative *cis*- (Figure 1, a) or the relative *trans*-configuration (Figure 1, b). In case of **5h**, however, the *trans*-configured isomer (Figure 2, b) dominates the *cis*-iso-

Table 1. Ce^{III}-catalyzed reaction of dicarbonyls 1 and styrenes 4 with atmospheric oxygen as oxidant

Donor	Olefin	Hydroperoxide Yield (%)										
		5	R	X	Y	n	Yield (%)	dr	6		2	
1a	4a	a	Н	CH ₂	OEt	0	68	62:38	a	4	a	3
1a	4b	b	Me	CH_2	OEt	0	87	64:36	_	_	a	5
1b	4a	c	Н	CH_2	OEt	1	50	50:50	c	1	b	25
1b	4b	d	Me	CH_2	OEt	1	42	100	_	_	b	25
1c	4a	e	H	CH_2	OMe	2	54	38:35:27	e	n. d. ^[a]	c	46
1c	4b	f	Me	CH_2	OMe	2	53	38:25:23:14	_	_	c	17
1d	4a	g	H	0	Me	_	64	54:46	g	8	d	18
1d	4b	ĥ	Me	O	Me	_	73	71:10:19	_	_	d	n. d. ^[a]
1e	4a	i	Н	NBn	Me	_	5	100	i	45	e	n. d. ^[a]
1e	4b	j	Me	NBn	Me	_	52	60:40	_	_	e	n. d. ^[a]

[a] n. d. = not detected.

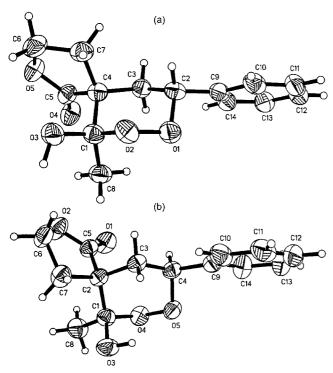


Figure 1. ORTEP view of the diastereoisomers of hydroperoxide **5g**; (a) relative *cis*-configuration and (b) relative *trans*-configuration of the lactone and phenyl group

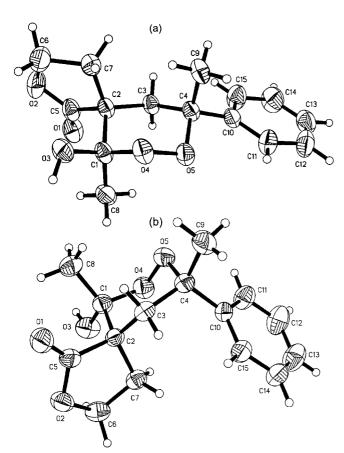


Figure 2. ORTEP view of the diastereoisomers of hydroperoxide **5h**; (a) relative *cis*-configuration and (b) relative *trans*-configuration of the lactone and phenyl group

mer (Figure 2, a) (trans:cis = 7:1). Also, the crystallographic data of the two diastereomers of $\mathbf{5c}$ confirm the cis/ trans configuration, but unfortunately, due to low crystal quality in the case of the cis-isomer, the data cannot be published. The remarkably high selectivity in the reversible formation of the hemiacetalic, anomeric centers can be explained by stereoelectronic effects, which are common in carbohydrate chemistry, and by the presence of intramolecular hydrogen bonds.

The transformation of the epimeric mixtures of the endoperoxides 5 to unique products is, of course, a fundamental issue. As shown in Scheme 3 for peroxide 5a, we succeeded in converting the styrene-derived products 5 by treatment with acetyl chloride (AcCl)/pyridine in dichloromethane (path b). An excess of acetyl chloride mediates the disproportionation of the peroxo function in 5a to give ketone 6a, which was isolated in 78 % yield.

$$CO_2Et$$
 + 4a (a) CO_2Et (b) CO_2Et (b) CO_2Et (b) CO_2Et (b) (c) $(c$

Scheme 3. Preparation of ketone **6a** from **1a** via the diastereomeric mixture of hydroperoxide **5a**; reagents and conditions: (a) **4a** (2 equiv.), CeCl₃·7H₂O (5 mol %), *i*PrOH, 23 °C, air; (b) AcCl (2 equiv.), pyridine (2 equiv.), CH₂Cl₂, 23 °C, 20 h

Conclusion

C–C coupling reactions are fundamental to the construction of carbon skeletons. We report, for the first time, a synthetically applicable oxidative coupling of β-dicarbonyl compounds 1 with olefins 4. In contrast to other known procedures that require stoichiometric amounts of cerium, manganese or cobalt salts, the presented conversion proceeds under catalysis with the inexpensive and non-toxic cerium salt CeCl₃·7H₂O, using atmospheric oxygen as oxidant at ambient temperature. The mixtures of diastereoisomeric hydroperoxides 5 obtained disproportionate to give dioxo esters 6, in a reaction mediated by acetyl chloride/pyridine. Dioxo derivatives 6 are of significant synthetic interest, due to their 1,4-dicarbonyl structural motif, whose synthesis usually requires the application of an umpolung strategy.

Experimental Section

General: The following compound was prepared according to literature procedures: **1e**.^[1b] All other starting materials are commercially available. Column chromatography was carried out using Merck SiO₂ 60 with hexanes (PE, b.p. 40–60 °C), and EtOAc (EA) as eluents. ¹³C NMR multiplicities were determined with DEPT experiments.

General Procedure for the Cerium-Catalyzed Hydroperoxide Formation: The respective substrate 1 (1 equiv.) was added to a mixture of CeCl₃·7H₂O (0.05 equiv.) and the respective olefin 4 (2 equiv.) in *i*PrOH (ca. 0.5 mL/1.6 mmol 1), and the reaction mixture was

stirred at 23 °C for 16 h under atmospheric oxygen. After removal of all volatile materials under vacuum, the residue was filtered through or chromatographed on SiO₂ (PE/EA) to yield the products 5.

Ethyl 1-Hydroxy-4-phenyl-2,3-dioxabicyclo[4.3.0]nonane-6-carboxylate (5a): According to the General Procedure, 1a (249 mg, 1.59 mmol), CeCl₃·7H₂O (30 mg, 0.081 mmol), and **4a** (334 mg, 3.21 mmol) in iPrOH (0.5 mL) were converted. Filtration through a short column of SiO₂ (PE:EA, 10:1) gave a crude fraction consisting of diastereomers A and B of 5a in a ratio of 62:38 (315 mg, 1.08 mmol, 68 %), alcohol 2a (7 mg, 0.04 mmol, 3 %), and ketone 6a (17 mg, 0.062 mmol, 4 %). By repeated chromatography (SiO₂, PE:EA, 5:1), by-products **2a** ($R_f = 0.25$) and **6a** ($R_f = 0.22$), and diastereomer A ($R_f = 0.19$) and B ($R_f = 0.26$) were completely separated. Diastereomer A: ¹H NMR (500 MHz, CDCl₃): $\delta = 1.29$ $(t, J = 7.1 \text{ Hz}, 3 \text{ H}, \text{CH}_3), 1.75 - 1.86 \text{ (m, 1 H, CH₂)}, 1.88 - 2.02$ (m, 3 H, CH₂), 2.17–2.27 (m, 2 H, CH₂), 2.30–2.37 (m, 1 H, CH₂), $2.67 \text{ (dd, } J = 11.9, J = 14.3 \text{ Hz}, 1 \text{ H, CH}_2), 4.15 \text{ (br. s, 1 H, OH)},$ 4.18-4.25 (m, 2 H, CH₂), 5.26 (dd, J = 2.5, J = 11.9 Hz, 1 H, CH), 7.32-7.40 (m, 5 H, CH) ppm. ¹³C{¹H} NMR (63 MHz, CDCl₃): $\delta = 14.06$ (CH₃), 19.30 (CH₂), 31.61 (CH₂), 32.50 (CH₂), 35.13 (CH₂), 52.69 (C), 61.43 (CH₂), 78.58 (CH), 108.37 (C), 127.08 (CH), 128.60 (CH), 128.85 (CH), 137.43 (C), 174.57 (C) ppm. IR (film): $\tilde{v} = 3352$ (br. s), 1708 (s), 1255 (m), 1177 (m), 1160 (m), 1154 (m), 1102 (m), 1044 (m) cm⁻¹. MS (FAB, glycerol): m/z (%) = 293 (75) $[MH^+]$, 275 (74) $[M^+ - H_2O]$, 259 (82) $[M^+ - H_2O_2]$, 229 (40), 201 (100), 153 (64), 105 (15). $C_{16}H_{20}O_5$ (292.33): calcd. C65.74, H 6.90; found C 65.52, H 6.90. **Diastereomer B:** ¹H NMR (500 MHz, CDCl₃): $\delta = 1.38$ (t, J = 7.3 Hz, 3 H, CH₃), 1.74–1.89 (m, 3 H, CH₂), 1.97-2.01 (m, 1 H, CH₂), 1.97 (dd, J = 10.7, J =14.1 Hz, 1 H, CH₂), 2.24-2.36 (m, 2 H, CH₂), 2.55 (dd, J = 5.2, $J = 14.1 \text{ Hz}, 1 \text{ H}, \text{CH}_2$, $4.20-4.35 \text{ (m, 2 H, CH}_2$), 5.15 (dd, J =5.2, J = 10.7 Hz, 1 H, CH), 6.20 (s, 1 H, OH), 7.29 - 7.40 (m, 5 H, CH) ppm. ${}^{13}C\{{}^{1}H\}$ NMR (63 MHz, CDCl₃): $\delta = 14.23$ (CH₃), 20.01 (CH₂), 34.95 (CH₂), 35.54 (CH₂), 37.18 (CH₂), 51.74 (C), 61.77 (CH₂), 79.50 (CH), 111.07 (C), 125.98 (CH), 128.19 (CH), 128.67 (CH), 140.38 (C), 175.72 (C) ppm. C₁₆H₂₀O₅ (292.33): calcd. C 65.74, H 6.90; found C 65.59, H 6.94. Ketone 6a: ¹H NMR (300 MHz, CDCl₃): $\delta = 1.24$ (t, J = 7.1 Hz, 3 H), 2.03-2.17 (m, 3 H), 2.53-2.69 (m, 3 H), 3.48 (d, J = 18.6 Hz, 1 H), 3.86 (d, J =18.6 Hz, 1 H), 4.17 (q, J = 7.1 Hz, 2 H), 7.43-7.49 (m, 2 H), 7.55-7.60 (m, 1 H), 7.93-7.97 (m, 2 H) ppm. ${}^{13}C\{{}^{1}H\}$ NMR $(125 \text{ MHz}, \text{CDCl}_3)$: $\delta = 14.00 \text{ (CH}_3)$, $19.86 \text{ (CH}_2)$, $33.40 \text{ (CH}_2)$, 37.76 (CH₂), 43.48 (CH₂), 57.47 (C), 61.66 (CH₂), 128.11 (CH), 128.65 (CH), 133.47 (CH), 136.31 (C), 170.71 (C), 196.74 (C), 215.11 (C) ppm. IR (film): $\tilde{v} = 2980$ (m), 1755 (s), 1721 (s), 1684 (s), 1451 (m), 1401 (m), 1359 (m), 1285 (m), 1224 (s), 1186 (m), 1153 (m), 1105 (m), 1039 (m), 1001 (m) cm $^{-1}$. MS (CI, CH₄): m/z $(\%) = 274 (34) [M^+], 257 (21), 229 [M^+ - OEt], 201 [M^+ -$ CO₂Et], 183 (5), 155 (17), 105 (53), 91 (7), 77 (8). HRMS calcd. for C₁₆H₁₈O₄: 274.1205; found 274.1205 [M⁺]. C₁₆H₁₈O₄ (274.31): calcd. C 70.06, H 6.61; found C 69.57, H 6.62. The data of 2a were in accordance with the literature.[1b]

Ethyl 1-Hydroxy-4-methyl-4-phenyl-2,3-dioxabicyclo[4.3.0]nonane-6-carboxylate (5b): According to the General Procedure, 1a (254 mg, 1.63 mmol), CeCl₃·7H₂O (30 mg, 0.081 mmol), and 4b (386 mg, 3.27 mmol) in *i*PrOH (0.5 mL) were converted. Filtration through SiO₂ and subsequent chromatography (SiO₂, PE:EA, 5:1) gave three fractions. The first fraction contained diastereomer A of 5b ($R_{\rm f} = 0.38, 281$ mg, 0.917 mmol, 56 %) as colorless crystals, m.p. 66 °C. The second fraction contained diastereomer B of 5b ($R_{\rm f} = 0.23, 153$ mg, 0.499 mmol, 31 %) as a colorless oil. In the third

fraction, alcohol 2a was obtained ($R_f = 0.17$, 14 mg, 0.081 mmol, 5 %). Diastereomer A: ¹H NMR (300 MHz, CDCl₃): $\delta = 0.91$ (t, $J = 7.2 \text{ Hz}, 3 \text{ H}, \text{ CH}_3), 1.57 \text{ (s, 3 H, CH}_3), 1.77-1.92 \text{ (m, 3 H)},$ 1.95-2.13 (m, 2 H), 2.11 (d, J = 14.0 Hz, 1 H), 2.20-2.30 (m, 1 H), 2.81 (d, J = 14.0 Hz, 1 H), 3.19 (dq, J = 10.8, J = 7.2 Hz, 1 H), 3.70 (dq, J = 10.7, J = 7.2 Hz, 1 H), 7.18-7.23 (m, 1 H, Ph),7.25-7.35 (m, 5 H, Ph and OH) ppm. ¹³C{¹H} NMR (63 MHz, CDCl₃): $\delta = 13.50$ (CH₃), 20.08 (CH₂), 30.03 (CH₃), 35.50 (CH₂), 36.10 (CH₂), 42.80 (CH₂), 49.61 (C), 61.00 (CH₂), 81.36 (C), 111.56 (C), 124.93 (CH), 126.95 (CH), 127.98 (CH), 143.53 (C), 176.33 (C) ppm. IR (ATR): $\tilde{v} = 3328$ (br. s), 2968 (m), 1692 (s), 1443 (s), 1369 (m), 1339 (m), 1300 (m), 1203 (m), 1113 (m) cm⁻¹. MS (FAB, glycerol): m/z (%) = 307 (61) [MH⁺], 289 (25) [MH⁺ - H₂O], 271 (38), 243 (31), 215 (100) $[MH^+ - H_2O - CO_2Et]$, 195 (10), 147 (5), 121 (14), 73 (22). C₁₇H₂₂O₅ (306.35): calcd. C 66.65, H 7.24; found C 66.63, H 7.24. Diastereomer B: ¹H NMR (300 MHz, CDCl₃): $\delta = 1.36$ (t, J = 7.0 Hz, 3 H, CH₃), 1.40 (s, 3 H, CH₃), 1.44–1.55 (m, 1 H), 1.65–1.77 (m, 2 H), 1.85–1.92 (m, 1 H), 1.97-2.03 (m, 1 H), 2.06-2.13 (m, 1 H), 2.39 (d, J = 14.0 Hz, 1 H), 2.70 (d, J = 14.0 Hz, 1 H), 4.24–4.34 (m, 2 H), 6.06 (br. s, 1 H), 7.26-7.30 (m, 1 H), 7.36-7.39 (m, 2 H), 7.46-7.47 (m, 2 H) ppm. ${}^{13}C\{{}^{1}H\}$ NMR (63 MHz, CDCl₃): $\delta = 14.14$ (CH₃), 19.95 (CH₂), 29.19 (CH₃), 34.59 (CH₂), 36.15 (CH₂), 38.90 (CH₂), 51.18 (C), 61.52 (CH₂), 81.03 (C), 109.89 (C), 124.69 (CH), 126.87 (CH), 128.32 (CH), 146.41 (C), 176.47 (C) ppm. C₁₇H₂₂O₅ (306.35): calcd. C 66.65, H 7.24; found C 66.50, H 7.19. The data of 2a were in accordance with the literature.[1b]

Ethyl 1-Hydroxy-4-phenyl-2,3-dioxabicyclo[4.4.0]decane-6-carboxylate (5c): According to the General Procedure, 1b (267 mg, 1.57 mmol), CeCl₃·7H₂O (30 mg, 0.081 mmol), and **4a** (331 mg, 3.18 mmol) in iPrOH (0.5 mL) were converted. Filtration through a short column of SiO₂ (PE:EA, 10:1) gave a crude fraction consisting of diastereomers A and B of 5c in a ratio 1:1 (240 mg, 0.785 mmol, 50 %), alcohol 2b (68 mg, 0.39 mmol, 25 %) and ketone 6c (5 mg, 0.02 mmol, 1 %). By chromatography (SiO₂, PE:EA, 5:1), by-products **2b** ($R_f = 0.19$) and **6c** ($R_f = 0.23$) were removed, and diastereomer A ($R_f = 0.32$) and B ($R_f = 0.15$) were separated completely and obtained as colorless solids. Diastereomer A: M.p. 118 °C. ¹H NMR (500 MHz, CDCl₃): $\delta = 1.35$ (t, J = 7.0 Hz, 3 H, CH₃), 1.51-1.65 (m, 2 H, CH₂), 1.71-1.82 (m, 4 H, CH₂), 2.03 $(dt, J = 13.4, J = 4.8 \text{ Hz}, 1 \text{ H}, CH_2), 2.43 (dd, J = 2.4, J =$ 14.2 Hz, 1 H, CH₂), 2.45-2.55 (m, 1 H, CH₂), 2.55 (dd, J = 11.7, $J = 14.2 \text{ Hz}, 1 \text{ H}, \text{ CH}_2$, $4.24 - 4.31 \text{ (m, 1 H, CH}_2$), $4.34 - 4.40 \text{ (m, 1 H, CH}_2)$, $4.34 - 4.40 \text{ (m, 1 H, CH}_2$), 1 H, CH₂), 5.13 (dd, J = 2.4, J = 11.6 Hz, 1 H, CH), 7.25–7.26 (m, 1 H, CH), 7.32–7.39 (m, 5 H, CH, OH) ppm. ¹³C{¹H} NMR (63 MHz, CDCl₃): $\delta = 14.10$ (CH₃), 20.18 (CH₂), 22.57 (CH₂), 30.45 (CH₂), 33.46 (CH₂), 35.76 (CH₂), 48.44 (C), 61.99 (CH₂), 81.51 (CH), 102.79 (C), 126.68 (CH), 128.63 (CH), 128.72 (CH), 137.54 (C), 177.71 (C) ppm. IR (ATR): $\tilde{v} = 3361$ (br. s), 2931 (m), 1706 (s), 1446 (m), 1278 (m), 1238 (s), 1174 (s), 1085 (m), 1025 (m), 1001 (m) cm⁻¹. MS (FAB, glycerol): m/z (%) = 307 (16) [M⁺], 289 (57) [M⁺ - H₂O], 273 (100), 243 (6), 215 (16), 199 (22), 195 (48), 170 (6), 123 (12), 105 (15), 75 (5). C₁₇H₂₂O₅ (306.15): calcd. C 66.65, H 7.24; found C 66.79, H 7.15. Diastereomer B: ¹H NMR (500 MHz, CDCl₃): $\delta = 1.28$ (t, J = 7.1 Hz, 3 H, CH₃), 1.47–1.57 (m, 2 H, CH₂), 1.63–1.69 (m, 2 H, CH₂), 1.82–1.93 (m, 3 H, CH₂), 2.23-2.31 (m, 2 H, CH₂), 2.65 (dd, J = 11.9, J = 14.0 Hz, 1 H, CH_2), 4.15-4.25 (m, 3 H, CH_2 , OH), 5.40 (dd, J = 2.4, J =11.9 Hz, 1 H, CH), 7.31-7.39 (m, 5 H, CH) ppm. ¹³C{¹H} NMR (63 MHz, CDCl₃): $\delta = 14.11$ (CH₃), 22.02 (CH₂), 22.54 (CH₂), 31.11 (CH₂), 34.06 (CH₂), 35.82 (CH₂), 48.31 (C), 61.22 (CH₂), 78.53 (CH), 99.30 (C), 127.12 (CH), 128.59 (CH), 128.88 (CH), 137.58 (C), 174.71 (C) ppm. C₁₇H₂₂O₅ (306.15): calcd. C 66.65, H

7.24; found C 66.47, H 7.18. **Ketone 6c:** M.p. 92 °C. ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$: $\delta = 1.27 \text{ (t, } J = 7.1 \text{ Hz, } 3 \text{ H, CH}_3), 1.74 - 1.88$ (m, 4 H, CH₂), 2.03-2.11 (m, 1 H, CH₂), 2.44-2.49 (m, 1 H, CH₂), 2.51-2.56 (m, 1 H, CH₂), 2.81-2.88 (m, 1 H, CH₂), 3.39 (d, J =17.4 Hz, 1 H, C*H*H), 3.57 (d, J = 17.4 Hz, 1 H, CH*H*), 4.19–4.29 (m, 2 H, CH₂), 7.43-7.47 (m, 2 H, CH), 7.54-7.58 (m, 1 H, CH), 7.94-7.97 (m, 2 H, CH) ppm. ¹³C{¹H} NMR (63 MHz, CDCl₃): $\delta = 14.05 \text{ (CH}_3), 22.03 \text{ (CH}_2), 26.82 \text{ (CH}_2), 36.74 \text{ (CH}_2), 40.55$ (CH₂), 44.03 (CH₂), 58.92 (C), 61.55 (CH₂), 128.06 (CH), 128.54 (CH), 133.12 (CH), 136.83 (C), 172.05 (C), 196.99 (C), 207.34 (C) ppm. IR (KBr): $\tilde{v} = 2968$ (m), 2958 (m), 2933 (m), 2866 (m), 1720 (s), 1677 (s), 1455 (m), 1387 (m), 1369 (m), 1313 (m), 1225 (m), 1183 (s), 1137 (m), 1071 (m), 1034 (m) cm⁻¹. MS (EI, 70 eV): m/z $(\%) = 288 (5) [M^+], 270 (2) [M^+ - H_2O], 243 (23) [M^+ - OEt],$ $215 (45) [M^+ - CO_2Et], 169 (100), 137 (21), 123 (40), 109 (15), 81$ (18), 77 (80). HRMS calcd. for C₁₇H₂₀O₄: 289.1435 [MH⁺]; found 289.1440. The data of **2b** were in accordance with the literature.^[1b]

Ethyl 1-Hydroxy-4-methyl-4-phenyl-2,3-dioxabicyclo[4.4.0]decane-**6-carboxylate (5d):** According to the General Procedure, **1b** (90 mg, 0.53 mmol), CeCl₃·7H₂O (10 mg, 0.027 mmol), and **4b** (125 mg, 1.06 mmol) in iPrOH (0.2 mL) were converted. By chromatography (SiO₂, PE:EA, 5:1), alcohol **2b** ($R_f = 0.19, 25 \%$) was separated, and 5d was obtained as colorless crystals ($R_{\rm f} = 0.36$, 69 mg, 0.22 mmol, 42 %), m.p. 121 °C. ¹H NMR (500 MHz, CDCl₃): δ = 1.07 (br. t, J = 6.4 Hz, 3 H, CH₃), 1.41 (br. s, 3 H, CH₃), 1.57–1.64 (m, 2 H), 1.66-1.82 (m, 5 H), 2.36-2.51 (m, 1 H), 2.56 (br. d, J =14.1 Hz, 1 H), 2.83 (d, J = 14.3 Hz, 1 H), 3.09-3.16 (m, 1 H), 3.66-3.73 (m, 1 H), 6.98 (br. s, 1 H, OH), 7.18-7.22 (m, 1 H), 7.29-7.32 (m, 2 H), 7.46-7.48 (m, 2 H) ppm. ${}^{13}C\{{}^{1}H\}$ NMR (63 MHz, CDCl₃): $\delta = 13.42$ (CH₃), 20.37 (CH₂), 22.34 (CH₂), 30.39 (CH₃), 30.65 (CH₂), 33.48 (CH₂), 39.73 (CH₂), 45.82 (C), 60.83 (CH₂), 82.30 (C), 102.24 (C), 126.10 (CH), 126.73 (CH), 127.56 (CH), 142.69 (C), 177.03 (C) ppm. IR (ATR): $\tilde{v} = 3316$ (br. s), 2937 (m), 1688 (s), 1456 (s), 1444 (s), 1370 (m), 1324 (m), 1265 (m), 1204 (s), 1186 (s), 1172 (s), 1095 (m) cm⁻¹. MS (FAB, glycerol): m/z (%) = 321 (58) [MH⁺], 303 (18) [MH⁺ - H₂O], 287 (67), 269 (7), 229 (9), 209 (100), 185 (25), 170 (14), 132 (17), 105 (24). C₁₈H₂₄O₅ (320.38): calcd. C 67.48, H 7.55; found C 67.54, H 7.52.

Methyl 7-Hydroxy-10-phenyl-8,9-dioxabicyclo[5.4.0]undecane-1-carboxylate (5e): According to the General Procedure, 1c (272 mg, 1.60 mmol), CeCl₃·7H₂O (30 mg, 0.081 mmol), and **4a** (336 mg, 3.23 mmol) in iPrOH (0.5 mL) were converted. Chromatography (SiO₂, PE:EA, 10:1) gave a crude fraction consisting of diastereomers A, B and C of 5e in a ratio of 38:35:27 (264 mg, 0.862 mmol, 54 %) and alcohol **2c** (137 mg, 0.735 mmol, 46 %). By repeated chromatography (SiO₂, PE:EA, 10:1) 2c was separated, and two product-containing fractions were obtained. The first fraction contained diastereomers A and B $[R_f (SiO_2, PE:EA, 5:1)] =$ 0.32] as a colorless oil. The second fraction contained diastereomer $C[R_f(SiO_2, PE:EA, 5:1) = 0.16]$ as a colorless oil. **Diastereomers A and B:** ¹H NMR (500 MHz, CDCl₃): $\delta = 1.43-1.58$ (m, 5 H, CH₂), 1.59-1.69 (m, 4 H, CH₂), 1.70-1.86 (m, 5 H, CH₂), 1.88-1.97 (m, 2 H, CH₂), 2.03 (ddd, J = 12.4, J = 5.3, J = 1.88-1.9714.3 Hz, 1 H, CH₂, isomer B), 2.11-2.16 (m, 2 H, CH₂), 2.21-2.30 (m, 2 H, CH₂), 2.36 (dd, J = 12.2, J = 14.0 Hz, 1 H, CH₂, isomerB), 2.40-2.44 (m, 1 H, CH₂, isomer B), 2.56 (dd, J = 4.0, J =14.0 Hz, 1 H, CH₂, isomer A), 3.26 (br. s, 1 H, OH, isomer B), 3.79 (s, 3 H, CH₃, isomer B), 3.84 (s, 3 H, CH₃, isomer A), 5.08 (dd, J = 2.1, J = 12.2 Hz, 1 H, CH, isomer B), 5.09 (dd, J = 4.0,J = 10.9 Hz, 1 H, CH, isomer A, 7.29 - 7.40 (m, 10 H, CH), 7.70(br. s, 1 H, OH, isomer A) ppm. IR (ATR): $\tilde{v} = 3338$ (br. s), 2930 (s), 2862 (m), 1734 (s), 1699 (s), 1453 (s), 1212 (s), 1110 (m), 1017 (s) cm⁻¹. MS (FAB, glycerol): m/z (%) = 307 (18) [MH⁺], 289 (64) $[MH^{+} - H_{2}O]$, 273 (100), 229 (2), 213 (25), 195 (14), 170 (2), 105 (24). C₁₇H₂₂O₅ (306.15): calcd. C 66.65, H 7.24; found C 66.57, H 7.20. **Diastereomer A:** $^{13}C\{^{1}H\}$ NMR (125 MHz, CDCl₃): $\delta =$ 21.98 (CH₂), 23.04 (CH₂), 24.94 (CH₂), 25.97 (CH₂), 38.13 (CH₂), 38.97 (CH₂), 51.60 (C), 52.97 (CH₃), 80.80 (CH), 102.88 (C), 126.39 (CH), 128.35 (CH), 128.57 (CH), 136.97 (C), 178.36 (C) ppm. Dia**stereomer B:** ${}^{13}C\{{}^{1}H\}$ NMR (125 MHz, CDCl₃): $\delta = 23.04$ (CH₂), 24.94 (CH₂), 25.97 (CH₂), 34.30 (CH₂), 37.56 (CH₂), 38.97 (CH₂), 51.60 (C), 51.92 (CH₃), 79.68 (CH), 102.88 (C), 127.17 (CH), 128.60 (CH), 128.95 (CH), 140.68 (C), 173.82 (C) ppm. Dia**stereomer C:** ¹H NMR (500 MHz, CDCl₃): $\delta = 1.45-1.68$ (m, 4) H, CH₂), 1.69-1.82 (m, 2 H, CH₂), 1.86 (ddd, J = 2.1, J = 6.8, $J = 14.4 \text{ Hz}, 1 \text{ H}, \text{ CH}_2$, 1.94–1.98 (m, 1 H, CH₂), 2.01 (dd, J =2.6, J = 13.8 Hz, 1 H, CH_2), 2.35-2.40 (m, 1 H, CH_2), 2.45-2.51(m, 1 H, CH₂), 2.64 (dd, J = 11.4, J = 13.8 Hz, 1 H, CH₂), 3.68 (s, 3 H, CH₃), 4.24 (br. s, 1 H, OH), 5.37 (dd, J = 2.6, J = 11.4 Hz, 1 H, CH), 7.29-7.42 (m, 5 H, CH) ppm. ¹³C{¹H} NMR (125 MHz, CDCl₃): $\delta = 19.40$ (CH₂), 22.33 (CH₂), 25.48 (CH₂), 26.91 (CH₂), 37.87 (CH₂), 38.70 (CH₂), 51.85 (CH₃), 52.00 (C), 78.55 (CH), 101.93 (C), 126.94 (CH), 128.56 (CH), 128.77 (CH), 137.69 (C), 175.39 (C) ppm. The data of 2c were in accordance with the literature.[1b]

Methyl 7-Hydroxy-10-methyl-10-phenyl-8,9-dioxabicyclo[5.4.0]undecane-1-carboxylate (5f): According to the General Procedure, 1c (291 mg, 1.71 mmol), CeCl₃·7H₂O (30 mg, 0.081 mmol), and 4b (393 mg, 3.33 mmol) in iPrOH (0.5 mL) were converted. Filtration through a short column of SiO₂ (PE:EA, 10:1) gave a crude fraction consisting of diastereomers A, B, C and D of 5f in a ratio 38:25:23:14 (288 mg, 0.899 mmol, 53 %) and alcohol 2c (54 mg, 0.290 mmol, 17 %). Alcohol 2c was separated by repeated chromatography (SiO₂, PE:EA, 10:1), and two product-containing fractions were obtained. The first fraction contained diastereomers A and D [R_f (PE/EA, 5:1) = 0.28, A:D, 80:20] as a colorless oil, and the second fraction contained diastereomers B and C [R_f (PE/EA, 5:1)= 0.34-0.44, B:C, 40:60] as a colorless solid. **Diastereomers A** and D: ¹H NMR (500 MHz, CDCl₃): $\delta = 1.16-1.28$ (m, 2 H), 1.31 (s, 3 H, CH₃, isomer A), 1.43-1.60 (m, 5 H), 1.61 (s, 3 H, CH₃, isomer D), 1.63-1.70 (m, 1 H), 1.70-1.87 (m, 8 H), 1.87-2.00 (m, 2 H), 2.12 (dd, J = 14.6, J = 8.9 Hz, 1 H, isomer A), 2.23 (d, J =14.0 Hz, 1 H, isomer D), 2.24 (d, J = 14.0 Hz, 1 H, isomer A), 2.43 (d, J = 14.6 Hz, 1 H, isomer D), 2.68-2.70 (m, 1 H, isomer D), 2.70 (d, J = 14.0 Hz, 1 H, isomer A), 3.39 (s, 1 H, OH, isomer D), 3.79 (s, 3 H, CH₃, isomer D), 3.79 (s, 3 H, CH₃, isomer A), 7.23-7.25 (m, 3 H, CH, isomer D), 7.28-7.30 (m, 2 H, CH, isomer D), 7.33-7.36 (m, 3 H, CH, isomer A), 7.40-7.41 (m, 2 H, CH, isomer A), 7.46 (br. s, 1 H, OH, isomer A) ppm. Diastereomer A: ¹³C{¹H} NMR (75 MHz, CDCl₃): $\delta = 21.63$ (CH₂), 22.76 (CH₂), 27.54 (CH₃), 28.94 (CH₂), 33.48 (CH₂), 37.07 (CH₂), 43.84 (CH₂), 50.65 (C), 52.27 (CH₃), 81.49 (C), 105.40 (C), 124.38 (CH), 126.76 (CH), 128.23 (CH), 147.21 (C), 178.34 (C) ppm. **Diastereomer D:** ¹³C{¹H} NMR (75 MHz, CDCl₃): $\delta = 23.45$ (CH₃), 23.60 (CH₂), 24.85 (CH₂), 26.48 (CH₂), 36.00 (CH₂), 37.74 (CH₂), 43.63 (CH₂), 50.37 (C), 51.75 (CH₃), 80.29 (C), 102.95 (C), 124.11 (CH), 127.66 (CH), 128.45 (CH), 145.06 (C), 174.80 (C) ppm. Diastereomers B and C: M.p. 124-129 °C. ¹H NMR (500 MHz, CDCl₃): $\delta =$ 1.22-1.37 (m, 2 H), 1.38 (s, 3 H, CH₃, isomer B), 1.39-1.51 (m, 2 H), 1.53-1.63 (m, 3 H), 1.57 (s, 3 H, CH₃, isomer C), 1.67-1.75 (m, 4 H), 1.77–1.82 (m, 5 H), 1.95–2.08 (m, 3 H), 2.12–2.22 (m, 2 H), 2.27 (d, J = 14.3 Hz, 1 H, CH₂, isomer C), 2.33–2.39 (m, 1 H), 2.69 (d, J = 14.4 Hz, 1 H, CH₂, isomer C), 2.83 (br. s, 1 H, CH₃, OH, isomer B), 2.83 (br. s, 3 H, CH₃, isomer B), 3.03 (s, 3

H, CH₃, isomer C), 3.35 (s, 1 H, OH, isomer C), 7.20-7.26 (m, 2 H, CH), 7.33-7.37 (m, 6 H, CH), 7.46-7.44 (m, 2 H, CH) ppm. IR (ATR): $\tilde{v} = 3464$ (br. m), 3329 (br. m), 2924 (m), 1696 (s), 1441 (m), 1336 (m), 1215 (m) cm⁻¹. MS (FAB, glycerol): m/z (%) = 321 $(45) [MH^+], 303 (32) [MH^+ - H_2O], 271 (8), [MH^+ - H_2O - O_2],$ 243 (18), 287 (70), 243 (18), 209 (100), 177 (52), 105 (24). C₁₈H₂₄O₅ (320.38): calcd. C 67.48, H 7.55; found C 67.28, H 7.55. Dia**stereomer B:** ${}^{13}C{}^{1}H}$ NMR (75 MHz, CDCl₃): $\delta = 22.31$ (CH₂), 22.51 (CH₂), 30.05 (CH₂), 30.31 (CH₃), 32.11 (CH₂), 38.12 (CH₂), 45.12 (CH₂), 49.27 (C), 51.20 (CH₃), 81.84 (C), 107.04 (C), 124.69 (CH), 126.53 (CH), 127.90 (CH), 143.88 (C), 178.12 (C) ppm. Dia**stereomer C:** ${}^{13}C\{{}^{1}H\}$ NMR (75 MHz, CDCl₃): $\delta = 23.29$ (CH₂), 24.87 (CH₂), 26.02 (CH₂), 30.97 (CH₃), 35.26 (CH₂), 37.91 (CH₂), 42.43 (CH₂), 49.36 (C), 50.80 (CH₃), 81.54 (C), 102.50 (C), 125.60 (CH), 126.64 (CH), 127.79 (CH), 142.53 (C), 172.80 (C) ppm. The data of 2c were in accordance with the literature.[1b]

6- Hydroxy-6-methyl-9-phenyl-2, 7, 8-triox aspiro [4.5] decan-1-one(5g): According to the General Procedure, 1d (197 mg, 1.54 mmol), CeCl₃·7H₂O (30 mg, 0.081 mmol) and 4a (342 mg, 3.28 mmol) in iPrOH (0.5 mL) were converted. Chromatography (SiO₂, PE:EA, 2:1) gave two fractions. The first fraction contained diastereomer A of 5g (130 mg, 0.492 mmol, 32 %) and ketone 6g (30 mg, 0.12 mmol, 8 %). The second fraction contained diastereomer B of 5g (129 mg, 0.488 mmol, 32 %) and alcohol 2d (39 mg, 0.27 mmol, 18 %). By repeated chromatography (SiO₂, PE:EA, 2:1) diastereoisomer A ($R_f = 0.25$) and B ($R_f = 0.13$) could be separated completely. Diastereomer A of 5g (cis): M.p. 145 °C. ¹H NMR (500 MHz, CDCl₃): $\delta = 1.90$ (s, 3 H), 1.90 (dd, J = 14.1, J =2.6 Hz, 1 H), 2.21 (ddd, J = 13.0, J = 9.7, J = 9.7 Hz, 1 H), 2.79 (dd, J = 14.1, J = 12.2 Hz, 1 H), 2.98 (s, 1 H), 3.15 (ddd, J =13.0, J = 7.5, J = 2.5 Hz, 1 H), 4.34-4.38 (m, 1 H), 4.43-4.48 (m, 1 H), 5.12 (dd, J = 12.3, J = 2.6 Hz, 1 H), 7.34–7.41 (m, 5 H) ppm. ${}^{13}C\{{}^{1}H\}$ NMR (125 MHz, CDCl₃): $\delta = 22.56$ (CH₃), 31.34 (CH₂), 39.67 (CH₂), 49.52 (C), 67.40 (CH₂), 80.21 (CH), 102.21 (C), 127.30 (CH), 128.79 (CH), 129.30 (CH), 135.92 (C), 177.46 (C) ppm. MS (FAB): m/z (%) = 265 (6) [MH⁺], 246 (70) $[M^+ - H_2O]$, 231 (100), 189 (12), 133 (33), 109 (30), 99 (17), 75 (76). C₁₄H₁₆O₅ (264.27): calcd. C 63.63, H 6.10; found C 63.68, H 6.19. Diastereomer B of 5g (trans): ¹H NMR (500 MHz, CDCl₃): $\delta = 1.49$ (s, 3 H), 2.12 (ddd, J = 14.1, J = 8.0, J = 5.7 Hz, 1 H), 2.15 (dd, J = 14.4, J = 2.2 Hz, 1 H), 2.40 (dd, J = 14.0, J = 14.0)11.8 Hz, 1 H), 2.73 (ddd, J = 8.7, J = 7.7, J = 14.0 Hz, 1 H), 3.49 (s, 1 H), 4.25-4.39 (m, 2 H), 5.80 (dd, J = 11.7, J = 2.3 Hz, 1 H), 7.32–7.39 (m, 5 H) ppm. ${}^{13}C\{{}^{1}H\}$ NMR (CDCl₃, 125 MHz): $\delta =$ 21.26 (CH₃), 33.13 (CH₂), 35.53 (CH₂), 47.11 (C), 64.85 (CH₂), 78.51 (CH), 99.68 (C), 127.12 (CH), 128.75 (CH), 129.15 (CH), 136.75 (C), 176.72 (C=O) ppm. **Ketone 6g:** ¹H NMR (500 MHz, CDCl₃): $\delta = 2.15$ (t, J = 13.2 Hz, 1 H), 2.29 (s, 3 H), 3.12 (ddd, J = 13.1, J = 5.7, J = 3.2 Hz, 1 H), 3.59 (d, J = 18.6 Hz, 1 H), 4.00 (d, J = 18.5 Hz, 1 H), 4.42 - 4.52 (m, 2 H), 7.48 - 7.52 (m, 2H), 7.61-7.64 (m, 1 H), 7.96-7.98 (m, 2 H) ppm. ${}^{13}C{}^{1}H{}^{1}$ NMR (125 MHz, CDCl₃): $\delta = 25.70$ (CH₃), 31.34 (CH₂), 44.20 (CH₂), 58.32 (C), 67.14 (CH₂), 128.16 (CH), 128.90 (CH), 134.10 (CH), 135.62 (C), 175.18 (C=O), 196.09 (C=O), 201.00 (C=O) ppm. MS (CI): m/z (%) = 247 (52) [MH⁺], 229 (67), 205 (30), 185 (100), 169 (1), 127 (11), 105 (41), 97 (5), 51 (5). The data of 2d were in accordance with the literature.[1b]

6-Hydroxy-6,9-dimethyl-9-phenyl-2,7,8-trioxaspiro[4.5]decan-1-one (5h): According to the General Procedure, **1d** (202 mg, 1.58 mmol), CeCl₃·7H₂O (30 mg, 0.081 mmol), and **4b** (379 mg, 3.21 mmol) in *i*PrOH (0.5 mL) were converted. Chromatography (SiO₂, PE:EA, 5:1) gave two fractions. The first fraction contained diastereoisomer

A and B of **5h** in a ratio 7:1 ($R_f = 0.22 - 0.40$, 260 mg, 0.934 mmol, 59 %) as a colorless solid. The second fraction contained diastereoisomer C of **5h** ($R_f = 0.15$, 63 mg, 0.23 mmol, 14%) as a colorless solid. Diastereomers A and B: M.p. 140 °C. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.33$ (s, 3 H, CH₃, isomer B), 1.46 (s, 3 H, CH_3 , isomer A), 1.47 (s, 3 H, CH_3 , isomer B), 1.54 (ddd, J = 9.8, J = 9.9, J = 13.6 Hz, 1 H, CH₂, isomer A), 1.67–1.75 (m, 2 H, CH₂, isomer B), 1.81 (s, 3 H, CH₃, isomer A), 2.10-2.16 (m, 1 H, CH_2 , isomer A), 2.25 (d, J = 14.2 Hz, 1 H, CHH, isomer B), 2.39 (d, J = 14.3 Hz, 1 H, CHH, isomer A), 2.73 (d, J = 14.3 Hz, 1 H,CHH, isomer A), 2.83 (d, J = 14.2 Hz, 1 H, CHH, isomer B), 3.62 (s, 1 H, OH, isomer A), 3.99-4.07 (m, 2 H, CH₂), 4.12-4.22 (m, 2 H, CH₂), 4.39 (s, 1 H, OH, isomer B), 7.23-7.30 (m, 2 H, CH), 7.33-7.40 (m, 4 H, CH), 7.45-7.50 (m, 4 H, CH) ppm. IR (ATR): $\tilde{v} = 3400$ (br. s), 1756 (m), 1735 (s), 1447 (m), 1381 (s), 1224 (s), 1194 (m), 1179 (m), 1029 (s) cm⁻¹. MS (FAB, glycerol): m/z (%) = $279 (50) [MH^{+}], 261 (30) [MH^{+} - H_{2}O], 245 (100), 243 (71) [MH^{+}]$ $-2 \text{ H}_2\text{O}$], 227 (16), 183 (18), 154 (26), 123 (26), 99 (33). $C_{15}H_{18}O_5$ (278.30): calcd. C 64.74, H 6.52; found C 64.35, H 6.49. Diastereomer A: ${}^{13}C\{{}^{1}H\}$ NMR (75 MHz, CDCl₃): $\delta = 22.31$ (CH₃), 29.91 (CH₃), 31.30 (CH₂), 42.70 (CH₂), 48.81 (C), 67.82 (CH₂), 81.61 (C), 101.87 (C), 125.18 (CH), 126.96 (CH), 128.20 (CH), 143.37 (C), 178.19 (C) ppm. **Diastereomer B:** ¹³C{¹H} NMR (75 MHz, CDCl₃): $\delta = 21.62$ (CH₃), 30.43 (CH₃), 31.49 (CH₂), 38.91 (CH₂), 47.14 (C), 66.12 (CH₂), 80.86 (C), 99.14 (C), 125.34 (CH), 127.10 (CH), 128.32 (CH), 142.94 (C), 177.99 (C) ppm. Diastereomer C: ¹H NMR (500 MHz, CDCl₃): $\delta = 1.57$ (br. s, 6 H, CH_3), 2.13 (br. s, 1 H, CH_2), 2.27 (ddd, J = 9.3, J = 9.3, J = 9.312.9 Hz, 1 H, CH₂), 2.74-2.79 (m, 1 H, CH₂), 3.05 (s, 1 H, OH), $3.15 \text{ (ddd, } J = 2.3, J = 6.7, J = 12.9 \text{ Hz}, 1 \text{ H, CH}_2), 4.33 \text{ (dt, } J =$ 2.3, J = 8.7 Hz, 1 H, CH_2), 4.38-4.43 (m, 1 H, CH_2), 7.26-7.30(m, 1 H, CH), 7.34-7.39 (m, 4 H, CH) ppm. ¹³C{¹H} NMR $(75 \text{ MHz}, \text{CDCl}_3)$: $\delta = 21.44 \text{ (CH}_3), 22.67 \text{ (CH}_3), 33.90 \text{ (CH}_2),$ 40.96 (CH₂), 47.48 (C), 67.28 (CH₂), 79.56 (C), 99.39 (C), 124.34 (CH), 128.39 (CH), 128.54 (CH), 143.68 (CH), 177.14 (CH) ppm.

2-Benzyl-6-hydroxy-6-methyl-9-phenyl-7,8-dioxa-2-azaspiro[4.5]decan-1-one (5i): According to the General Procedure, 1e (332 mg, 1.53 mmol), CeCl₃·7H₂O (30 mg, 0.081 mmol), and **4a** (337 mg, 3.24 mmol) in iPrOH (0.5 mL) were converted. Chromatography (SiO₂, PE:EA, 5:1) gave two fractions. The first fraction contained ketone **6i** [R_f (PE:EA, 2:1) = 0.17, 229 mg, 0.683 mmol, 45%] as a yellow oil. In the second fraction, 5i was obtained $[R_f]$ (PE:EA, 2:1) = 0.14, 28 mg, 0.079 mmol, 5 %] as a colorless solid, m.p. 171 °C. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.30$ (s, 3 H, CH₃), 1.63 (dd, J = 2.1, J = 13.5 Hz, 1 H, CH₂), 2.15 (ddd, J = 8.5, J = 5.8, J = 5.814.0 Hz, 1 H, CHH), 2.42 (ddd, J = 8.3, J = 5.7, J = 13.9 Hz, 1 H, CHH), 2.80-2.88 (m, 1 H, CH₂), 3.21-3.35 (m, 2 H, CH₂), 4.42 (d, J = 14.4 Hz, 1 H, CHH), 4.59 (d, J = 14.4 Hz, 1 H, CHH),5.15 (dd, J = 2.0, J = 12.0 Hz, 1 H, CH), 5.85 (s, 1 H, OH),7.22-7.25 (m, 2 H, CH), 7.30-7.42 (m, 8 H, CH) ppm. ${}^{13}C\{{}^{1}H\}$ NMR (75 MHz, CDCl₃): $\delta = 22.31$ (CH₃), 27.37 (CH₂), 35.77 (CH₂), 43.89 (CH₂), 47.21 (CH₂), 48.71 (C), 78.71 (CH), 100.25 (C), 127.44 (CH), 128.10 (CH), 128.28 (CH), 128.62 (CH), 128.95 (CH), 129.07 (CH), 135.48 (C), 136.95 (C), 175.64 (C) ppm. IR (ATR): $\tilde{v} = 3227$ (br. s), 2951 (m), 1651 (s), 1494 (m), 1451 (m), 1290 (m), 1244 (m), 1124 (m), 1097 (s) cm⁻¹. MS (FAB): m/z (%) = $354 (32) [MH^{+}], 336 (61) [MH^{+} - H₂O], 320 (24), 296 (11), 275$ (15), 245 (12), 215 (16), 188 (26), 183 (32), 91 (100). HRMS calcd. for C₂₁H₂₃NO₄: 354.1705 [MH⁺]; found 354.1724. C₂₁H₂₃NO₄ (353.16): calcd. C 71.37, H 6.56, N 3.96; found C 71.19, H 6.54, N 3.19. **Ketone 6i:** ¹H NMR (300 MHz, CDCl₃): $\delta = 1.84$ (ddd, J =7.8, J = 9.6, J = 13.4 Hz, 1 H, CH₂), 2.25 (s, 3 H, CH₃), 2.92 (ddd, J = 2.7, J = 8.2, J = 13.4 Hz, 1 H, CH₂), 3.25 (ddd, J = 9.6, J = 13.4 Hz, 1 H, CH₂), 3.25 (ddd, J = 13.4 Hz, 1 Hz,

Table 2. X-ray crystallographic data for 5g and 5h

	cis-5g	trans-5g	cis-5h	trans-5h
Empirical formula	$C_{14}H_{16}O_{5}$	$C_{14}H_{16}O_{5}$	C ₁₅ H ₁₈ O ₅	C ₁₅ H ₁₈ O ₅
Formula mass [g mol ⁻¹]	264.27	264.27	278.28	278.28
Crystal system	monoclinic	monoclinic	monoclinic	monoclinic
Space group	$P2_1/c$	$P2_1/c$	$P2_1/c$	$P2_1/c$
a (Å)	10.9452(6)	6.5947(2)	8.1593(9)	8.3476(12)
b (Å)	11.6229(8)	15.5436(6)	10.7326(8)	16.156(3)
$c(\mathring{A})$	10.3322(9)	12.8034(9)	15.6899(11)	10.6008(13)
β [°]	102.948(6)	102.690(5)	92.689(7)	103.217(10)
Z	4	4	4	4
$V(\mathring{\mathbf{A}}^3)$	1280.99(16)	1280.36(11)	1372.5(2)	1391.8(3)
$\delta_{\text{calcd.}} [\text{g cm}^{-3}]$	1.370	1.371	1.347	1.328
$R_{\rm w}(F^2)$	0.1703	0.1573	0.1567	0.0991
$R(F)$ $[I > 2\sigma(I)]$	0.0639	0.0560	0.0513	0.0582

2.7, J=9.7 Hz, 1 H, CH₂), 3.43 (ddd, J=8.0, J=8.1, J=12.6 Hz, 1 H, CH₂), 3.46 (d, J=18.8 Hz, 1 H, CH₂), 4.12 (d, J=18.7 Hz, 1 H, CH₂), 4.44–4.54 (m, 2 H, CH₂), 7.20–7.24 (m, 2 H, CH), 7.27–7.39 (m, 3 H, CH), 7.46–7.51 (m, 2 H, CH), 7.57–7.63 (m, 1 H, CH), 7.97–7.98 (m, 1 H, CH), 7.99–8.01 (m, 1 H, CH) ppm. 13 C{ 1 H} NMR (63 MHz, CDCl₃): $\delta=26.11$ (CH₃), 27.78 (CH₂), 44.60 (CH₂), 44.71 (CH₂), 47.31 (CH₂), 60.21 (C), 127.77 (CH), 128.06 (CH), 128.13 (CH), 128.76 (CH), 128.81 (CH), 133.70 (CH), 135.87 (C), 136.18 (C), 171.56 (C), 197.20 (C), 203.95 (C) ppm. IR (ATR): $\tilde{v}=1712$ (m), 1679 (s), 1449 (m), 1430 (m), 1260 (m), 1215 (m) cm⁻¹. MS (EI, 70 eV): m/z (%) = 335 (45) [M⁺], 292 (20) [M⁺ — COMe], 215 (43) [M⁺ — COMe — Ph], 188 (99), 105 (34), 91 (100). HRMS calcd. for $C_{21}H_{21}NO_3$: 335.1521 [M⁺]; found 335.1521.

2-Benzyl-6-hydroxy-6,9-dimethyl-9-phenyl-7,8-dioxa-2-azaspiro-[4.5]decan-1-one (5j): According to the General Procedure, 1e (126 mg, 0.580 mmol), CeCl₃·7H₂O (10 mg, 0.027 mmol), and 4b (126 mg, 1.06 mmol) in iPrOH (0.2 mL) were converted. Chromatography (SiO₂, PE:EA, 2:1) gave a fraction containing diastereomers A and B of 5j in a ratio 60:40 ($R_f = 0.19$, 110 mg, 0.299 mmol, 52 %) as a colorless solid, m.p. 144 °C. ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$: $\delta = 1.15 \text{ (s, 3 H, CH}_3, \text{ isomer B)}, 1.24-1.29$ (m, 1 H, CH₂, isomer A), 1.40–1.46 (m, 1 H, CH₂, isomer B), 1.48 (s, 3 H, CH₃, isomer A), 1.49 (s, 3 H, CH₃, isomer A), 1.76–1.77 (m, 1 H, CH₂, isomer A), 1.88 (s, 3 H, CH₃, isomer A), 2.19 (d, J = 14.0 Hz, 1 H, CHH, isomer B, 2.35 (d, <math>J = 14.6 Hz, 1 H,CHH, isomer A), 2.81-2.87 (m, 2 H, CH₂), 2.85 (d, J = 14.0 Hz, 1 H, CHH), 2.93-2.98 (m, 1 H, CH₂, isomer B), 3.00-3.05 (m, 1 H, CH₂, isomer B), 3.17–3.22 (m, 1 H, CH₂, isomer A), 3.32–3.37 (m, 1 H, CH₂, isomer A), 4.38 (d, J = 14.0 Hz, 1 H, CHH, isomer B), 4.42 (d, $J = 14.6 \,\text{Hz}$, 1 H, CHH, isomer A), 4.54 (d, J =14.6 Hz, 1 H, CHH, isomer A), 4.59 (d, J = 14.0 Hz, 1 H, CHH, isomer B), 5.54 (s, 1 H, OH, isomer B), 7.22-7.23 (m, 2 H, CH), 7.25-7.30 (m, 5 H, CH, OH), 7.30-7.34 (m, 2 H, CH), 7.34-7.39 (m, 8 H, CH), 7.50-7.53 (m, 4 H, CH) ppm. IR (ATR): $\tilde{v} = 3317$ (br. s), 1670 (s), 1495 (m), 1453 (m), 1444 (m), 1212 (m), 1178 (m), 1126 (m), 1067 (m) cm⁻¹. C₂₂H₂₅NO₄ (367.43): calcd. C 71.91, H 6.86, N 3.81; found C 72.18, H 6.69, N 3.73. Diastereomer A: ¹³C{¹H} NMR (63 MHz, CDCl₃): $\delta = 22.65$ (CH₃), 28.37 (CH₂), 30.23 (CH₃), 42.78 (CH₂), 45.43 (CH₂), 46.85 (CH₂), 50.62 (C), 81.68 (C), 103.00 (C), 125.28 (CH), 126.58 (CH), 127.36 (CH), 127.83 (CH), 127.98 (CH), 128.54 (CH), 136.36 (C), 144.89 (C), 174.13 (C) ppm. **Diastereomer B:** ¹³C{¹H} NMR (63 MHz, CDCl₃): $\delta = 21.72$ (CH₃), 27.87 (CH₂), 30.75 (CH₃), 38.84 (CH₂), 44.41 (CH₂), 47.26 (CH₂), 50.35 (C), 80.50 (C), 99.81 (C), 125.46 (CH), 126.72 (CH), 128.02 (CH), 128.09 (CH), 128.37 (CH), 128.86 (CH), 135.52 (C), 143.89 (C), 175.56 (C) ppm.

Transformation of Hydroperoxide 5a into Dioxo Ester 6a: AcCl (100 mg, 1.3 mmol) was added at 0 °C to a solution of 5a (165 mg, 0.56 mmol) and pyridine (112 mg, 1.4 mmol) in CH₂Cl₂ (0.5 mL), and the reaction mixture was stirred at room temp. for 20 h. After removal of all volatile materials under vacuum, the residue was purified by chromatography on SiO₂ with PE/EA, 5:1 ($R_f = 0.22$) to give 6a as a colorless oil (121 mg, 0.44 mmol, 78 %). ¹H NMR (300 MHz, CDCl₃): $\delta = 1.24$ (t, J = 7.1 Hz, 3 H), 2.03-2.17 (m, 3 H), 2.53-2.69 (m, 3 H), 3.48 (d, J = 18.6 Hz, 1 H), 3.86 (d, J =18.6 Hz, 1 H), 4.17 (q, J = 7.1 Hz, 2 H), 7.43-7.49 (m, 2 H), 7.55-7.60 (m, 1 H), 7.93-7.97 (m, 2 H) ppm. ${}^{13}C{}^{1}H{}^{1}$ NMR (125 MHz, CDCl₃): $\delta = 14.00$ (CH₃), 19.86 (CH₂), 33.40 (CH₂), 37.76 (CH₂), 43.48 (CH₂), 57.47 (C), 61.66 (CH₂), 128.11 (CH), 128.65 (CH), 133.47 (CH), 136.31 (C), 170.71 (C), 196.74 (C), 215.11 (C) ppm. IR (film): $\tilde{v} = 1755$ (s), 1721 (s), 1684 (s), 1451 (m), 1401 (m), 1359 (m), 1285 (m), 1224 (s), 1186 (m), 1153 (m), 1105 (m), 1039 (m), 1001 (m) cm $^{-1}$. HRMS for $C_{16}H_{18}O_4$ (70 eV, EI): calcd. 274.1205, found 274.1205. C₁₆H₁₈O₄ (274.32): calcd. C 70.06, H 6.61; found C 69.57, H 6.62.

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